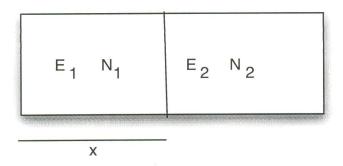
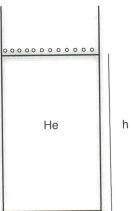
Date due: February 3, 2006

1. Consider a horizontal cylinder of length a, and cross section A, divided into two parts by a piston P of mass m. The piston can slide back and forth inside the cylinder without any friction. The walls of the cylinder as well as the piston are perfect heat insulators. The left side of the piston is filled with N_1 molecules of helium gas, and the right side is filled with N_2 molecules of Helium gas. The total energy of the system is fixed and is given by E. We assume that the piston is heavy and moves slowly so that at any instant of time we can regard the gas on each side to be in equilibrium, obeying ideal gas equation of state.



- a) Take the distance x of the piston from the left wall, the velocity v of the piston and the energy E_1 of the gas on the left side of the piston as independent variables and write down the equations of motion in the standard form with first order time derivatives only.
- b) Express the total entropy of the system in terms of the independent variables E_1 , x and v and show, using the equations of motion, that the total entropy does not change with time.
- c) Now assume that the system is not in complete thermal equilibrium during the motion of the piston so that the pressure on the piston due to the gas on left compartment differs from the one calculated using the ideal gas equation of state by c_1v and similarly pressure on the piston due to the gas on right compartment differs from the one calculated using the ideal gas equation of state by c_2v for some constants c_1 , c_2 . Find the equations of motion to first order in c_1 , c_2 .

- d) Calculate the rate of change of entropy of the system. Show that for physical choice of the sign of c_1 , c_2 the entropy increases with time.
- 2. Consider a vertical cylinder of cross section A containing helium gas, closed from the top by a piston P of zero mass. The piston can slide up and down without any friction. The walls of the cylinder as well as the piston are perfect heat insulators. The piston is held in place by putting 2n equal weights, each of mass m, on top of it. The piston is at a height h above the base of the cylinder.



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- a) Now suppose we suddenly remove n of the weights and wait till the system comes to rest. You can assume that the system is in equilibrium at the beginning and the end of the process but not in the intermediate stage. What will be the final height of the piston? What is the difference between the initial and the final entropy?
- b) Next consider the case where we remove n of the weights one by one, and after each removal wait for the system to come to rest before the next one is removed. What will be the height of the system at the end of the process? What is the difference between the initial and the final entropy?

- c) Study what happens in the $n \to \infty$ limit.
- 3. Consider a one dimensional harmonic oscillator with phase space coordinates q, p and Hamiltonian

$$H(q,p) = \frac{p^2}{2m} + \frac{1}{2}Cq^2$$
,

where m and C are constants. The system is in thermal contact with a reservoir at temperature T.

- a) Calculate the average energy of the oscillator.
- b) Calculate the probability that the energy of the oscillator lies between E and $E + \delta E$.
- c) Calculate the average value of q^2 .
- 4. Consider a system of N magnetic dipoles, each carrying a dipole moment μ . The dipoles are fixed in space at their centers but are free to rotate about any axis passing through their centers and perpendicular to the axis of the dipole. The moment of inertia of each of the dipoles about any such axis is I. The system is kept in thermal equilibrium with a heat bath of temperature T_R . There is also an external uniform magnetic field B which we take to be in the z direction.
 - a) Choose appropriate coordinates and momenta for describing this system and write down the Hamiltonian in terms of these coordinates and momenta.
 - b) Calculate the free energy F, the entropy S, mean energy E and the specific heat at constant magnetic field C_B of the system.
 - c) Show that the average magnetization of the system is given by $-\partial F/\partial B$. Find an expression for the average magnetization and also the susceptibility, which is defined as the rate of change of the average magnetization with respect to the external magnetic field B.
 - d) Estimate the total fluctuation in the average energy as well as the average magnetization in the canonical ensemble.

e) Now suppose that the dipoles are not fixed in their positions but are allowed to move freely inside a box of volume V. Determine how the results of a) - d) will change.

Use what (x +

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Jacobian of the transfur is

the same
$$\rightarrow$$

e.g. \rightarrow Take $1-D$ and (2) \not
 $p'=p-e/c$ $A(2)$
 $q'=2$
 $\therefore \partial p/p=1$
 $\partial p/q=-e/c$ A'

red particles moving in a uniform magnetic.

5. Consider a set of N free charged particles moving in a uniform magnetic field inside a box of size V. If \vec{A} is the vector potential so that $\vec{B} = \vec{\nabla} \times \vec{A}$ is the magnetic field, then the hamiltonian of the system is given by

$$H = \frac{1}{2m} \sum_{i=1}^{N} \left(\vec{p}_i - \frac{e}{c} \vec{A}(\vec{x}_i) \right)^2, \qquad (7.5)$$

where m and e denote the mass and charge of the particle and e is the velocity of light. As before, the average magnetization is defined to be $-\partial F/\partial B$ where $B = |\vec{B}|$ and F is the free energy.

- a) Show that the average magnetization of the system vanishes.
- b) Show that the same result holds when we add to the hamiltonian an interaction term of the form:

$$\sum_{\substack{i,j=1\\i\neq j}}^{N} V(\vec{r_i}, \vec{r_j})$$

- 6. Consider a system of N objects which are free to move inside a box of volume V. Each of these objects, in turn, is made of a pair of point masses, connected by a spring whose length in the absence of any external force is l_0 and whose spring constant is κ . The system is in thermal contact with a heat bath of temperature T_R . (You can ignore the mass of the spring during this calculation.)
 - a) Write down the expression for the Hamiltonian of the system in terms of appropriate generalized coordinates and momenta.
 - b) Find an expression for the number of objects in the full system for which the mean separation between pair of particles constituting the object lies between l and l+dl.
 - c) Find an expression for the number of point masses in the full system whose velocity ($|\vec{v}|$) lies between v and v + dv.
- 7. Consider a box containing ideal gas, moving with uniform velocity v_0 .
 - a) Begin with the known velocity distribution function in the rest frame of the box and transform this to the laboratory frame to find the fraction of particles whose velocity $(|\vec{v}|)$ lies between v and $v + \delta v$.

finete for - o, too limits dealing with a are close

D-B12 K (1.10)2

4 Also, we are already dealing with V+00 & only when you are close to the surface, surface effects will matter

b) Now try to find the velocity distribution of particles in the moving box in the following way. We treat the total momentum as a conserved charge and define the microcanonical ensemble as the one for which

$$\rho = \delta(H(\vec{q}, \vec{p}) - E_0) \, \delta^{(3)}(\vec{P}(\vec{q}, \vec{p}) - \vec{P}_0)$$

where E_0 and \vec{P}_0 denote the total energy and momentum of the system and $\vec{P}(q,p)$ is that particular function of the generalized coordinates and momenta of the system which give the total momentum vector. In the same way that we calculated the distribution of energy among the individual molecules of the system at rest, we can now calculate the distribution of energy and momentum among individual molecules of the moving system. Find the velocity distribution function in the moving system this way and compare this to the answer obtained in the previous analysis.

Now with this evaluate (Px7, (Py), (Px))

Rempare with the result expected.

We will find that 25R. P' is related to \$\overline{\partial}{\partial

- Canonical Ensemble A finite or large system in thermal contact with a nuch bigger system. of (\(\bar{q},\bar{p}\)=Ne-H(\bar{q},\bar{p})/k\(\bar{q}\) J d'e d' e - BH(Z), F) F(Z), F) < F(2, F) / TR = [d" 2 d" e - B H (\$7, F") where B= kTR Suppose a system is large.

2(E) increases rapidly with E.

Define (F) micro

Sang dup F (T, F) J (HEF) F)

(AN MAL SINE TILE) Sangamp S(H(B,F)-E) (upto a normalisation factor) (F (9, F)) = Num
Den Num = JdE Jdng dnp F(29F) e F(24 12, F) = = SdEe-BE (F) micro 12 (F) decreases smooth of E rapidly (with E) (ben faige)

Integral will be dominated by ENE. OF (-BE+ hor(E))=0 ₹S(E) => - 1 * te + t 1 * T(E) = 0 7 T(E)=TR Num ~ <F7 = PdE e-PE D(E) a smooth fr Den = Sate e PER(E) = (F(2, F)) te 2 the other is a sparply peaked fr.) (F) (F) $= \left\langle F(\overline{2}, \overline{\beta}) \right\rangle_{\overline{E}}^{micro}$ For large systems, Cana d' Microcan, ensemble will give the T(E)=TR same ans. for all averages E is calculated # Definition ly solving this Partition function

2(B)= N (dng dnp e-BH(Z,F) Lithis relation should be st found to relate microcan. defined for fixed (analog of st, which will appear in E, & canonicap den, but neverthelen ite not the same as I) defined for fixed $\Rightarrow 2(\beta) = N \int d^{n}q d^{n}b \int d\xi \delta(\xi - \mathcal{H}(\bar{x}, \bar{y}))$ Entropy can always be defined even for a small system = de e-BE 12 (E) but you can't do thermodyn, with it

 $\Rightarrow 2(\beta) = \int d\xi e^{-\beta\xi + \frac{1}{\lambda}S(\xi)}$ The integrand is sharply peaked book entry & & has maximum at E e S/k is --we do Paylors, around & series expansion around & $2(\beta) = \int dE e^{-\beta \vec{E} + \frac{1}{2} S(\vec{E})} dE$ $= e^{-\beta \bar{E} + \frac{1}{k} S(\bar{E})}$ $= e^{-\beta \bar{E} + \frac{1}{k} S(\bar{E})}$ The integrand Can have more than one local max, - then look at the B= ATE global max. 2 T | 225/2 | Definition Free energy P(P) = - x TR ln Z $= -k \operatorname{Tr} \left(-k$ ~ In N - Jean be ~ F-TR S(E) ignored S(E)= E-F En S, --- refer to quantities Calculated in canonial ensemble Romentin: (this E is indeed & boos a commical ensemble has TR as the input parameter & has no & otherwise) can any only of the minute with a finite scar of the start of the star Remembers F=F-TS t(E)=TR So we To F(Ty V) for a gas system by T.

at Nos Star. df = dE-tds-SdT of can be ., of = - b dv - SdT defined for microcani as S + 0f =- b) 2F =- S has been defined E= F+TS (Once we calculate f, using 2, everything it) H Example: 2 deal gas for finite system, The M mono atomic molecules gm 2 defin 2F6+ =-5 k 5= khur wm It agree - it depends = NV (2m #) 3M on your choice which defin you we 1. F= -kTln Z= - kTln N-MkTlnV wrnit be of much thermodyn use - morpe of an academic - 3MKT L (2mtht) exercise 1. S=-2F = & lm N + M & lm V Sh (2mtt kt) +] + 3M k Sh (2mtt kt) +] + 3M k of row how $E = F + TS = \frac{3M\lambda T}{2}$ p=- OF = MAT FOUCHAT 5 found here & 5 17 for eventle at the rame

- Pake me molecule of a gas as the finite system of the freat the gest of the gas as reservoir. Example 91, P1: phase space coordinates of
this molecule pre (7, 7)

(7, 7)

= 5dq, d37, e-pre (7, 7) < F(91, FD) = - (21, F) B = KT, temperature of the gas is (scor the rest of the gas is acting as the reservoir) Take P(2, 7) = O(H(2, 5,)-E) O(E+8E-)+(2, 7)) Ly Step function 0 for 20 0 In $F \le f(\overline{2}, F) \le E + \delta E$ O otherwise (F) gire molecule to the energy of the molecule to the lie beth E& E+&E

were (F,F) = Stanfah, e to Fin Fi O (Fron - F) D(E+&E-Form)

The control of the energy of the ene fraction of phone space Journal up
is prized up partied where $P_1 = P_1 = P_2 = P_3 = P_$

Take SE = small p² 7,2m € > p1>√2m€ PI (2m (E+SE) 7 P, (2m(E+SE) Name + 2 Jam SE Num: $(2mE)e^{-\frac{1}{2}mE}$ +0 $(8E^2)$ Num: $(2mE)e^{-\frac{1}{2}mE}$ $\sqrt{2mE}$ $\sqrt{2m}$ $\sqrt{2m}$ Probability distribution for gustrod of many evening Velocity distribution pythus in early singleme you are many the singleme you he wire P(v) 80= prob. of a molecule having velocity better v & v+80 with the gives and swell of sw The gas \$ () du=29 (E) 8 F Nothing fells we lead that the bank of four of four or the for a more many E=== mo2 => 8E=mo 80 of the grown = mv P(E) $= mv \cdot 2 = \beta^{3/2} (2mv^2)^{1/2}$ $\times e^{-1/2} \beta^{3/2}$ P(v) = mo P(E) lo was show hid

Microcanonical ensemble $P(\overline{2}, \overline{F}) = N \delta(\mathcal{H}(\overline{2}, \overline{F}) - E)$ Canonical ensemble f(q,F)=N=PH(q,F) Norm, factus cancel out imp, What can we pay about N? I focus on the canonical ensemble. for ideal gas S= &MhV+ 3Mk (h+1) + 3Mk h (2TTmk) + KluN(M) M: Potal no. of molecules faside the following experiment pv=MXT. S= &M ln (MAT) + 3MA (ln T+i) + 3M k ln (2 tmk) + k ln N (M), now of the particle Consider the following experiment! P, T, MI +, T, M2 Now remove the partition. 8. Should the entropy change ? ms. 7 9f S is extensive then it should not change,

(We'll get back the original situation if we put back the fartition). M1 2(1, T) + M2 s(1, +) Entropy shouldn't = (MI+MZ)s(b,T) VIMI V2 - inhially (coord, had small rol.) (V1+V2) MI+M2) finally (The way as have defined phase space vol. of as a result entropy, they change). Adjust N's so that the charge in N's odjusted by the charge in N's y i'mitially Vi MI V2 MINN(MI) I i'mitially Vi MI+M2 ... (V1+V2) MI+M2 (M+M2) - finally The source of the non-extensive prop. of formule for me entropy is the term KM lu (MKT) = tx lu (MM km Tm). S should grow ~ M] S(P, T, M1+M2) - S(B, T, M2) - S(B, T, M2) = 0 K(M1+M2) h(M1+M2) - KM1 hM1 - KM2hM2 f & h N(M1+M2)-kln N(M1)-kh N(M2) = 0 Depre f(M) = ln N(M) + M lm M Process looks indicates
quishable microscopicely f (M) + f (M2) [At the you frest the portides os macroscopic Tist. - Clars colly particles should be treated as inshable of sol. Jam's

for M= M, M=E f(M+E)-f(M)=f(E)=f(O) E =) E df &m = f (0) E f=f'(0)M+K40 C(M+H2) +K = CMI+K+CN2+K -) K=0 Solution f(M) = (M) constant In or (M) = CM - M In M = (M - (h m! +M) In 201 . for large M = ulnze-ze =) u lun = hu! +u = EM-In M! (where ~= (-1) 7 N(M) = EM M! Set c= 0 We sell set Consider 2-particle phase space = MT \ d^3/1 d^3/1 d^3/2 d^3/2 d For indistinguishable particles: And comeniary set & do (F₁ = F₁(0), F₁ = F₂(0), F₂ = F₂(0) is the same phase space foint as $(\vec{p}_1 = \vec{p}_2), \vec{p}_1 = \vec{q}_2), \vec{p}_2 = \vec{p}_1), \vec{q}_2 = \vec{q}_1(0)$ their the for M farkely we have MI

(We'll use this as an od hoe rule in class. Stat. mech. & that No AT.) [In order to treat S as the extensive prof.)

particles should be indistinguishable] (Don't have to mix up with my with dishinguishebility of particles - Don't have to absorb It into
the phase space integral - take it as the
norm. factor - so no conflict with class stat.
mech.) # 9f the gas contains a mixture of M, particles of type 1, M2 particles of type 2, etc. Then we can choose N= M, 1, M21, --- Mx1. A big system of (reservoir) in contact with a finite system S such that they can exchange energy as well as Eo = ER + E , energy of S Lenuys of R total energy Mo = MR + M > # of molecules

total # of molecules in R

molecules

no. of K(1, \$\bar{p}, M): Some for of the phase space coordinates (\$\bar{p}, \bar{p}) of M.

(K(2, F, M)) = (+ (2), p, m) can system (2+5) is total Joseph Co temp of the Mo! Sd 20 d 20 K (2, A, M) = BH(Q, F,M)-BH(Q, F,M)-BH(Q, F,M) MR) can fræget about
the res.
The hare at

Jeffined no - Mo! d 20 d 3Mob e- p) (2', F) - p) (2', F) S by itself is not to! Sd 3Mg to (2, F, M) e - 82 (0 (20, Fo, Mo) large ne cessarily Moi d3Mo to d3Mo to e PHO (EG, FO, MO) Numerator = 1 [Money] d 2 (2) F, M) e - Port (2), F, M) - Porter, Money We have faker prison ped so we must First int. Q. & F. now the man the farker on Sd3 90 = Sed399 + Sd391 = \(\frac{1}{M=0} \frac{1}{M} \) \(\frac{1}{M} Zp (B, M.-M)

Now, e - Po & Fr (Mo-M) - M 2 Mo + 2 M2 2 Fr (Pomo) - 2 Mo A Suneral or negligible for M - finite beas we are ML< Mo dealing with = P - Po FR (Po, Mo) + MR M a finite system when MR = Po OFR (Po, Mo) The way he \[
\left\{\frac{1}{2},\beta,m\right\}
\right\}
\\
\frac{1}{2} \tau officialing byte approvious popus some dere probable with some of the some of the sound of X Pore (Bo, Mo) are reging over an ensemble are ranging energy & ticles with varying no. of particles ranging no. of particles of \$(2), F) + 4, M, (2), F) + 4, M, (3), M, (4), M, (5), F) + 4, M, (5), F) + 4, M, (6), F) + 4 han come the or we start of the > goard canonical ensemble the forther disk refrished

< K(2, F, M) } can 20/1/06 = M=0 MI SK (7, F, M) d 2 d 3 M/b x e-pri(F) 3 M/k/m 3 No 1 Sd34 & d3 6 - Pola, 8, 11 + 4 km or from of a party A control of the cont mo to (\$,2, m) d3m q d3m pe Bo 24(\$,2, m) +4km

N50 M. (\$(2, p), m) d3m q d3m pe Bo 24(\$,2, m) +4km mmming our $\mathbb{R}(\overline{2},\overline{p},M)$ Mrs 60 om o Change convention

Change ForMo)

MR = OMO MGO IN SK(Q, F, M) 29 2 2 PEN (Q, F, M)+Bo, K, M (K(T, F, M)) Bo, Mo 1 SP-PO)+ (2, F), M) +MR POM 3M d3M -> Grand Canoni'Cal M=0 mil a finit ensemble infinité énury geservoir har In cononical, E was a parkides let so your for all possible values up to a by amoning the reservoir

 $\Rightarrow \langle \kappa(\vec{2}, \vec{p}, M) \rangle_{\beta_0, M_6}$ E POHRM (K) PO, M Z (PO, M) 2 e POHRM 2 (Po, M) Z(Po, M)
Suppose S is large.

t o Bo. MRM the defr. of 7 (Pg M) Expect & BOURM 2(BO, M) is sharply beaked F=-KThZ POMRM-POF (Po, M) 72= e = e = e Find

BOMR-BO OF (BO, M) = 6

maximum M(Bo, M) for of the sub-system => u(Po, M)= MR determines M determines M Can Deposite 2(\$0,9M) KY = CAN MODE 2(\$0,9M) M=0 POMRM Z(\$0,9M) No ster in the particles in the the rubo yothm but don't will carry drug from M

=> < K> grand
po, MR -> determined from Grand Canonical

partition function

partition of the M 2 (\$0,9 M)

S = 2 (\$0,9 M) (Bo, M) = MR large size POMRM Z (PO, M) BoMRM+MZ(Po, M) Jn 9 ~ = BOMRM-BOF (POOM) Answing of principle of the see to we see to the see to the see of Consider a large system loith temp. To foresmore popular Volume V, chemical footential un volume V, energy E, # of molecules M etc. other thermodyn, use to quartities arely sylver lu S= xt {uM-F} mere all arends using the samply whatly have been ply have been ply have been planted to be a sea of the same of the s dF=-SdT-pdV -> doesn't what harppens when # of molecules i, [dF=-sdt-pdv+µdm] : (0 F/m) T, V /L d(mg)=-dt suf-f) + ttlthow-walth

$$\Rightarrow d(\ln 9) = + \frac{d\tau}{k\tau^2} \left\{ -\mu M + F + \tau 3 \right\}^{\alpha} + \frac{M}{k\tau} d\mu + \frac{1}{k\tau} d\nu \right\}$$

$$\Rightarrow \left(\frac{\partial}{\partial \tau} \ln 9 \right)_{\mu, \nu} = \frac{1}{k\tau^2} \left(-\mu M + F + \tau 3 \right)$$

$$= \frac{1}{k\tau^2} \left(\frac{F}{\tau} - \mu M \right)$$

$$= \frac{1}$$

E- MM = KT2 & (m 8) = xt2 { - xt2 e pr v (2mt & t)3/2 + 3 PMV (2m th) 3/2 - 12/2 7 E-MM = -MM + & T2 x 32 e PM v (2mTh) 3h Th => E= 3 kt V (2m # kt)3heph 7 [E = 3 MAT Even, a physical situation dictales p= kt & h & a particular = LTE &M (2mt Lt) 3h ensemble, you can use any is see the right ensemble acc. provided to lease of The short front in calculation 7 TOVEMET Grand can: will come handy cohen you have to do some 9= Inisand dipepele, F, M) +pun counting 3m = pr((a, F)+Pum) = 1 5 m. (3m & 3m p = pr((a, F)+Pum) = 9 m. then no constraint get & every same on It of partidy all give parys will make calaulation for finite systems were corrections to of part-fr, les contraint = PM menns easier There or which is differ. Compare a du 9) T, V = M/k7 calculation end of enough

\$ h 9 = 1 39 = (- E + u m) exemple and quantity Relation of 11 to (Eggn) pis the response the were in expression dF=-Sdt-pdv+ugy "she of her hand d(E-TS)J. Jos. a list fricky 7 dE-tds-sdT =-sdt-pdv+udM 30 is not an org. 7 + ds = dE+pdv-udM 7 ds = + dE + P/A dv-ly dM => (as) am)E,V=-MT $\left(\frac{\partial S}{\partial M}\right)_{1} = \left(\frac{\partial S}{\partial M}\right)_{2}$ the same up to the Fricher of T 7 1/2 5 1/2 fluctuation in M.
for goand canonical
for ensemble But I has to match an 2 sides for 2 systems in eggin > 50 分かって M has to match an 2 Sog Mi gra < K(E, B, M) grand - 1. (can) Popul sides (Wot $\frac{\sum \{K\}_{\beta,\mu}^{can} \in \beta,M}{\sum \{\beta,M\}}$ $\sum \{K\}_{\beta,\mu}^{can} \in \{\beta,M\}$ $\sum \{K\}_{\beta,n}^{can} \in \beta,M - \beta F(\beta,M)$ $\sum \{B,n\} \in \{\beta,M\}$ $\sum \{B,n\} \in \{\beta,M\}$ writing volume factor 8y sterr

PMM- PF (8 9 M) = Q M M - Q F (\$, M) - (\$ $-\beta_{12} \frac{\partial^{2} F(\beta, \overline{m})}{\partial \overline{m}^{2}} (M - \overline{M})^{2} +$ > guarkity controlling the peak if zero - no peak peak if zero, 12F 2M2 if large - peak is sharp Calculation of 27F is extensine F(P, M, V) =M (P, M) 2 = f (P, V/M) + M &m f (P, V/M) = f(BVm) - v & f(P, Vm) f is a homogenery = f(B, V/M) - V & F(B, M,V) of degree zero = f(B) W+ 4 p(B) YM) b -> homegeneous - 3M f (P, V/M) V/M) - 1/2 p(P, V/M) - 1/18. In of degree Bero in M/V - intensine MT Zm p(P) V/m) = \frac{\frac{1}{2} \frac{1}{2} \frac\frac{1}{2} \frac{1}{2} \frac{1}{2} \frac{1}{2} \frac{1}{2} \frac + 1/2 (-V 3/p(P, VM))

Define: $f = \frac{M}{V}$, $V = \frac{1}{9} = V/M$ rolyme/ farticle = - 62 36 · M = 3t . I IM N Sp. 1 32F N NAT TOP 1 JOB 1 NAT 10861 VM ., AM small for large M provided 1 2/5/ 70 moter mer 3m2 3p68 mux pe xve provided many serving and serv in those cares servity from to space ty or or not you ther shows system the shows at show the sho

