

1/3/06

# Liquid-gas phase transition

$p$ : pressure,  $\rho = m/V = \text{density}$ ,  
 $T = \text{temperature}$ .

Equation of state: -  $p = g(\rho, T)$

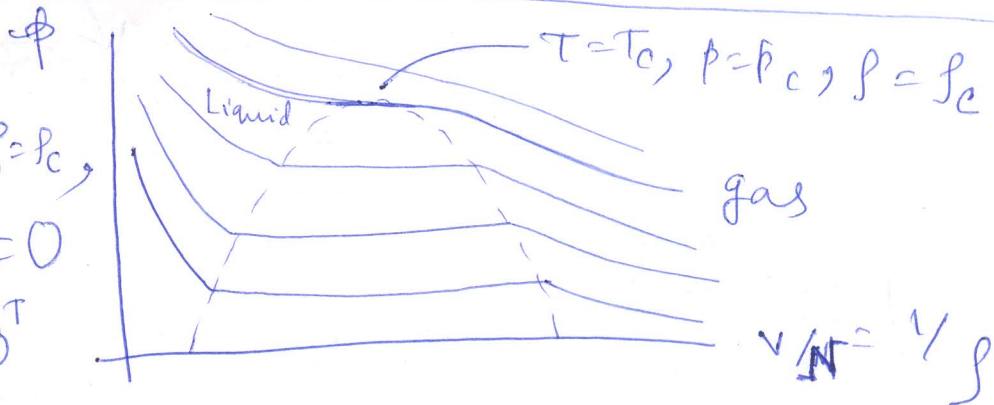
Relation betw.  $\rho$  &  $T$

↓  
some function

$$p/T = \frac{\partial \ln Q}{\partial V} = \frac{1}{V} \ln Q$$

Since  $\ln Q \propto V f(\rho, T)$

## Phase diagram for Liquid-gas system



At  $T=T_c, p=p_c, \rho=\rho_c$ ,

$$\left(\frac{\partial p}{\partial v}\right)_{N,T} = 0, \left(\frac{\partial^2 p}{\partial v^2}\right)_{N,T} = 0$$

$$\Rightarrow \frac{\partial p}{\partial \rho} = 0, \frac{\partial^2 p}{\partial \rho^2} = 0$$

Q) Can we understand this from Statistical mechanics?

$$p = p_c + a (\rho - \rho_c)^3 \quad \text{at } T=T_c.$$

$$\rho - \rho_c = \left(\frac{p - p_c}{a}\right)^{1/3}$$

$$\frac{\partial p}{\partial \rho} \propto (\rho - \rho_c)^{-2/3} \rightarrow \infty \text{ as } \rho \rightarrow \rho_c.$$

In order to understand phase transition from statistical mechanics, we need to understand the origin of non-analyticity.

There is some kind of non-analyticity in the  $p$  vs  $\rho$  det.  $p - p_c$  in terms of  $\rho - \rho_c$

[ we haven't gotten this diag. from stat. mech. Till now we have used only thermodyn. relations ]



$\Rightarrow f$  is discontinuous as a fcn of  $\beta$ .

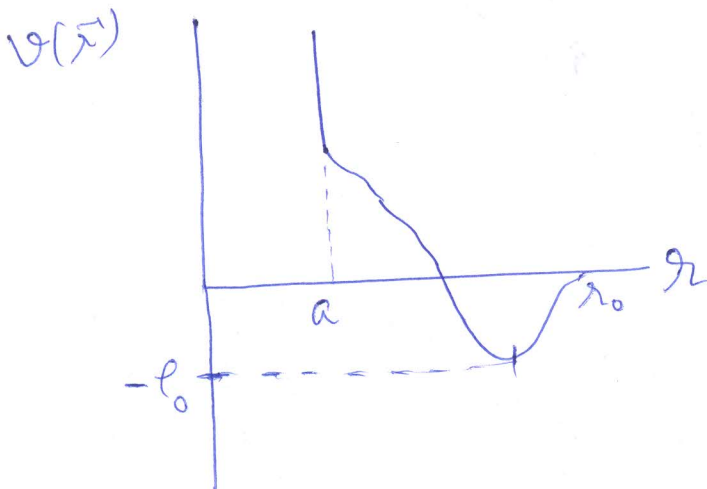
### Model of imperfect gas (classical)

$$H = \sum_{i=1}^N \frac{\vec{p}_i^2}{2m} + V(\vec{r}_1, \dots, \vec{r}_N)$$

$$\sum_{i < j} v(\vec{r}_i - \vec{r}_j)$$

### Assumptions

- ①  $v(\vec{r}) = \infty$  for  $|\vec{r}| \leq a$
- ②  $v(\vec{r}) \geq -\epsilon_0$  for  $a < |\vec{r}| < r_0$
- ③  $v(\vec{r}) = 0$  for  $r \geq r_0$



known as  
Hard sphere  
interaction

Hard core int.

Molecules are rep. for very near approach

It is possible that  $v$  never becomes  $-ve$

But it can't be arbitrarily large negative

$\epsilon_0$  can be the or  $-ve$

In the realistic case  $\epsilon_0$  should be positive bcs there is some attractive component

